AMENDMENTS TO THE SPECIFICATION

Please replace the name identifying the nitrogen-containing ligand structure "4,4',4"-tri-*tert*-butyl-2,2':6',2'-terpyridine" on the bottom left-hand corner of page 8 of the specification with the name:

-- 4,4',4"-tri-*tert*-butyl-2,2':6',2"-terpyridine --.

Please replace the paragraph describing Example 15 on pages 24-25 of the application with the following amended paragraph:

-- In an argon glovebox, a 30 mL septum bottle was loaded with a 0.0061 g quantity of $CuCl_2.2H_2O$ (FW 170.98, 0.0357 mmol) and 15 mL of toluene. Then, a 0.0130 g quantity of ~~4,4',4" tri-*tert* butyl-2,2':6',2' terpyridine~~ 4,4',4"-tri-*tert*-butyl-2,2':6',2"-terpyridine (FW 401.60, 0.0324 mmol) ligand was added resulting in the formation of a yellow solution. A 0.77 g quantity of 30 wt % MAO solution in toluene was then added. The yellow solution turned colorless upon MAO addition. Next, 5 g of *n*-butyl acrylate (FW 128.17, 0.039 mol) was added. The bottle was sealed in the glove box and placed in a fume hood. The solution was stirred at 25°C for 72 hours. The viscous solution was added to a MeOH/HCl (300 mL MeOH/100 mL 10% HCl) solution to precipitate the polymer. The product was washed with water, then methanol, and dried in vacuum oven at 60°C for 24 hours. The yield of the poly(*n*-butyl acrylate) was 3.12 g. The IR spectrum (film) of the product showed the characteristic polymer ester absorption peak at 1736 cm^{-1}. On polymerization, the monomeric ester absorption peak shifted from 1728 cm^{-1} to the polymeric ester absorption at 1736 cm^{-1}. The characteristic double bond absorption peaks at 1637 cm^{-1} and 812 cm^{-1} also disappeared upon polymerization. The GPC data (solvent: THF, polystyrene standard) gave a M_n of 124,100 and a M_w of 214,600 ^{13}C NMR (ppm,$CDCl_3$): 13.7 [s, $-CH_2-CH(COOCH_2CH_2CH_2\underline{C}H_3)-$], 19.1 [s,

-CH_2-CH ($COOCH_2CH_2\underline{C}H_2CH_3$)-], 30.7 [s, -$CH_2$-CH($COOCH_2\underline{C}H_2CH_2CH_3$)-], 34-37 [m, -$\underline{C}H_2CH(COOCH_2\,CH_2\,CH_2CH_3)$-], 41-42 [m, -$CH_2$-$\underline{C}H(COOCH_2CH_2CH_2CH_3)$-], 64.5 [s, -$CH_2$-CH($COO\underline{C}H_2CH_2CH_2CH_3$)-], 174-175 [m, -$CH_2$-CH($\underline{C}OOCH_2CH_2CH_2CH_3$)-]. There were no resonances due to olefin from the monomer. --